

## IN THE CLAIMS

Claims 1-26 (canceled)

Add the following claims:

-27 (currently amended). A process for obtaining polyglycolyl urea resin from aromatic diglycinates for insulating electric conductor, in the absence of HCN polluting residues, comprising the following steps:

A) preparing a methyl diglycinate:

(i)[a]) reacting a mixture of methylhaloester and methylenedianiline in the presence of C<sub>1</sub>—C<sub>4</sub> aliphatic solvent under reflux conditions at atmospheric pressure [and up to] at a solvent reflux temperature of 58 – 63°C, wherein said methylhaloester is selected from the group consisting of methylbromopropionate or methylchloropropionate;

(ii)[b]) adding triethylamine, [as catalyst] a rate of 0.178 l/hr. per Kg of reactants;

(iii)[c]) separating the solvent through atmospheric distillation [till] until 40% of its initial volume is recovered;

(iv)[d]) cooling [at] the reaction solution at 20 °C [understirring and beginning at 50°C] under stirring and then adding the drinking water at a volume adequate to

dissolve the bromine salt obtained;

(v) [e] filtering and purifying the diglycinate by washing with water;

(vi) [f] drying the methyl diglycinate obtained; and

B) preparing polyglycolyl urea resin:

(i)[a] stirring together a suspension of cresylic acid and said methyl diglycinate in  
a reactor at room temperature, stirring until a solution is formed;

(ii)[b] reacting the obtained diglycinate with aromatic isocyanate in the presence of  
[a solvent as] cresylic acid in a reactor until solution is complete at] adding  
methylene diisocyanate under constant stirring to said solution of said cresylic acid  
and methyl diglycinate, and keeping temperature of said solution from rising above  
60 °C;

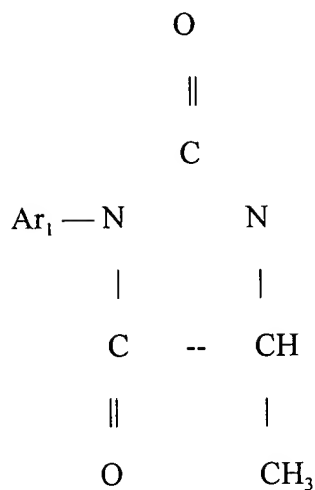
(iii)[b] reacting the diglycinate preferable with metilen diisocyanate solvent and  
catalyst at a temperature of 200°C] adding a catalyzer to said solution of ii);

(iv) raising the temperature of the solution up to 200° C.;

c) distilling and then cooling the reaction product; and

d) recovering the polyglycolyl urea resin having the formula I:

I



n

where Ar<sub>1</sub> is a substitute aromatic compound [such as a substitute diphenylalkyl], and [2 < n 500]  
2<n<500.

28. (canceled).

29. (currently amended) The process according to claim 27 wherein the mixture reflux is conducted for [at least 16] up to 19 hours

30. (canceled)

31. (canceled)

32. (currently amended) The process according to claim 27 wherein the resin obtained is cooled [at] to a temperature of 70°C

33. (currently amended) The process according to claim 27 wherein the catalyst in step B(iii) is selected from the group consisting of trethylenediamino octane and 1,4 diazobicyclo (2,2,2) octane.  
[and is added at temperatures up to 180 °C]

34.(currently amended.) The process according to claim 27 wherein the polyglycolyl urea resin obtained has viscosity (Cp) of 4,800 at 15% solids at 70°C.

35. (previously added) The process according to claim 27, wherein the C<sub>1</sub>—C<sub>4</sub> aliphatic is methanol.

36. (currently amended) The process according to claim 27, wherein the aromatic diglycinate is [preferable] a methyl diglycinate [obtained and is dried with hot air at 40°C and] that corresponds to a stereoisomer mixture [with] having a melting point of 95 – 116°C and having [of] the following formula II:



wherein Ar<sub>1</sub> represents aromatic rings.